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# Formal Synthesis of Premisakinolide A and C(19)−C(32) of Swinholide A via Site-Selective C−H Allylation and Crotylation of Unprotected Diols

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# **S** Supporting Information

[AB](#page-2-0)STRACT: [Using stereo-](#page-2-0) and site-selective C−H allylation and crotylation of unprotected diols, an intermediate in the synthesis of premisakinolide A (bistheonellic acid B) that was previously made in 16−27 (LLS) steps is now prepared in only nine steps. This fragment also represents a synthesis of C(19)−C(32) of the actin-binding macrodiolide swinholide A.



The efficacy of anticancer agents that perturb microtubule<br>dynamics (e.g., paclitaxel, docetaxel, ixabepilone, eribulin<br>movelato)<sup>1</sup> suggests astin binding agents hold promise for  $mesylate)^1$  suggests actin-binding agents hold promise for cancer therapy.<sup>2</sup> Swinholide  $A$ , an actin-binding marine polyketid[e](#page-2-0) isolated in 1985 from the Okinawan marine sponge Theonella swinh[oe](#page-2-0)i,<sup>3a</sup> was first assumed to be secreted by symbiotic cyanobacteria<sup>3c</sup> but was later shown using Palauan specimens of the [sp](#page-2-0)onge to be linked to the presence of hete[r](#page-2-0)otrophic unicellular bacteria.<sup>4</sup> Swinholide A was initially mistaken for the monomeric macrolide hemiswinholide  $A^{2a}$ until subs[e](#page-2-0)quent work $3b-e$  revealed it to be a 44-membered macrodiolide (Figure 1). Other members of this class have be[en](#page-2-0) isolated (swinholides B–[K,](#page-2-0) ankaraholides A and B),<sup>5</sup> including isoswinholide $5a$  and their presumed biogenic precursor preswinholide [\(swinho](#page-1-0)lide A seco acid).<sup>6</sup> Many ma[rin](#page-2-0)e natural products, the [m](#page-2-0)iskinolides (bistheonellides) $\alpha$  and scytophycin  $C_1^8$  have structural features in comm[on](#page-2-0) with swinholide A (Figure 1).

[S](#page-2-0)winholide A is the most potent member of its class, with [cytotoxicit](#page-1-0)y against diverse tumor cell lines in the ng/mL range<sup>9</sup> due to its ability to dimerize actin ( $K_d \approx 50$  nM)<sup>10a</sup> and cleave actin filaments.<sup>10</sup> High levels of potency are critically depe[nd](#page-2-0)ent on the symmetric 44-membered diolide [ring](#page-3-0) of swinholide A, as co[ng](#page-3-0)eners such as isoswinholide A are significantly less potent.<sup>9</sup> A high resolution  $(2 \text{ Å})$  crystal structure of swinholide A bound to two actin molecules has bee[n](#page-2-0) obtained,<sup>10c</sup> enabling the rational design of simplified actin-binding compounds.<sup>1</sup>

Total synthe[ses](#page-3-0) of swinholide A were reported by Paterson<sup>12</sup> and Nicolaou.<sup>13</sup> The tot[al](#page-3-0) synthesis of preswinholide A was reported by Nakata,<sup>14</sup> and numerous syntheses of swinholide [A](#page-3-0) substructures [ha](#page-3-0)ve been disclosed.<sup>15</sup> Broadly speaking, these syntheses all expl[oit](#page-3-0) classical carbonyl additions involving stoichiometric organometallic rea[gen](#page-3-0)ts. We have developed catalytic enantioselective methods that enable direct stereo- and site-selective conversion of lower alcohols to higher alcohols.<sup>16</sup>

These methods streamline or eliminate protecting group manipulations and discrete alcohol-to-carbonyl redox reactions, providing the most concise routes ever reported to diverse polyketide natural products.<sup>16b</sup> Here, we apply these methods to the construction of the  $C(19)-C(32)$  fragment of swinholide A and the for[mal](#page-3-0) synthesis of premisakinolide A (bistheonellic acid B), $^{15h}$  the monomer of the macrodiolide misakinolide A (bistheonellide A). $\frac{7}{2}$ 

Our retrosynthetic [ana](#page-3-0)lysis of premisakinolide A (bistheonellic acid B) and the  $C(19)-C(32)$  $C(19)-C(32)$  substructure of swinholide A is as follows (Scheme 1). Aldehydes 11 and 12 were envisioned to arise through cross-metathesis of vinyl pyran 5 with iodoether 8. [The syn](#page-1-0)thesis of vinyl pyran 5 takes advantage of site-selective C-allylation of (S)-1,3-butane diol 1.<sup>17</sup> The iodoether 8 is readily prepared from 2-methyl-1,3propane diol 6 via bidirectional enantioselective double antic[rot](#page-3-0)ylation.<sup>18</sup> The strategic advantage of these methods is borne out by the brevity our route, which delivers Miyashita's premisakin[oli](#page-3-0)de A (bistheonellic acid B) intermediate  $12^{15h}$  in nine steps, a substructure previously prepared in 16−27 steps.12b,i,15b,h

The synthesis of vinyl pyran 5 begins with the established stere[o- and s](#page-3-0)ite-selective allylation of commercially available  $(S)$ -1,3-butane diol 1.<sup>17</sup> Exposure of the resulting homoallylic alcohol 2 to the second generation Grubbs catalyst in the presence of cis-1,4-dia[ce](#page-3-0)toxy-2-butene delivered the product of cross metathesis 3 as a 5:1 mixture of alkene geometrical isomers.<sup>19</sup> Tsuji−Trost cyclization of 3 using the chiral palladium catalyst modified by the indicated chiral ligand delivere[d t](#page-3-0)he 2,6-trans-disubstituted pyran 4 in 90% yield as a 4:1 mixture of diastereomers. As previously established, the presence of alkene geometrical isomers at the stage of

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Figure 1. Swinholide A and selected naturally occurring congeners. For the structures of swinholides D−K, see ref 5.

Scheme 1. Retrosynthetic Analysis of Premisakinolide A (Bistheonellic Acid B) and  $C(19)-C(32)$  of Swinholide A and Prior Fragment Syntheses<sup>a</sup>



a For graphical summaries of prior total syntheses, see Supporting Information. Longest Linear Sequence (LLS). The carbon numbering scheme of swinholide A is used.

compound 3 does not influence diastereoselectivity in the cyclization to form pyran 4. While pyran 4 was previously prepared in our laboratory, $17$  the present route is an improved gram-scale synthesis with recycling of the iridium catalyst. Finally, methylation of th[e](#page-3-0) 4-hydroxyl moiety delivers vinyl pyran 5 (Scheme 2).

The PMB-protected iodoether 8 was prepared by a procedure previously established in our laboratory (Scheme  $3$ .<sup>18</sup> Specifically, diol 6 was subjected to double *anti*-crotylation to provide 7, which incorporates a triketide stereopolyad. A si[ngle](#page-3-0) enantiomer of 7 is formed due to Horeau's principle. $20$ That is, the minor diastereomer of the monoadduct is converted predominantly to t[he](#page-3-0) *meso-stereoisomer.*<sup>21</sup> In the conversion of diol 6 to adduct 7, it was found that use of  $\alpha$ methyl allyl acetate prepared from the correspondi[ng](#page-3-0) alcohol using acetic anhydride and triethylamine (rather than pyridine) gave optimal results. Iodoetherification differentiates the diol and terminal olefin moieties and defines the chirotopic nonstereogenic center at  $C(22)$ . Conversion of the remaining free hydroxyl to the PMB-ether delivers compound 8.

With vinyl pyran 5 and iodoether 8 in hand, conversion to aldehydes 11 and 12 (Miyashita's intermediate) was attempted. The cross-metathesis of vinyl pyran 5 and iodoether 8 was

#### Scheme 2. Synthesis of Vinyl Pyra[n](#page-2-0) 5 via Site-Selective Allylation of Diol  $1<sup>a</sup>$



a Cited yields are of diastereomeric mixtures isolated by silica gel chromatography. Diastereomeric ratios were determined by <sup>1</sup>H NMR analysis of crude reaction mixtures. See Supporting Information for further experimental details, including recovery and recycling of the iridium catalyst.

#### Scheme 3. Synthesis of Iodoether  $8^a$



a Cited yields are of diastereomeric mixtures isolated by silica gel chromatography. Diastereomeric ratios were determined by <sup>1</sup>H NMR analysis of crude reaction mixtures. See Supporting Information for further experimental details.

especially challenging due to competing isomerization of the terminal olefin of iodoether 8 to the internal olefin. Although related cross-metatheses of vinyl pyrans are known, $22$  they involve equatorially disposed vinyl moieties, unlike the vinyl moiety of pyran 5. After considerable optimization, it w[as f](#page-3-0)ound that use of the second generation Grubbs−Hoveyda catalyst in

<span id="page-2-0"></span>Scheme 4. Formal Synthesis of Premisakinolide A (Bistheonellic Acid B) and Synthesis of the C(19)−C(32) Substructure of Swinholide  $A^a$ 



 $^a$ Cited yields are of diastereomeric mixtures isolated by silica gel chromatography. Diastereomeric ratios were determined by  $^1{\rm H}$  NMR analysis of crude reaction mixtures. See Supporting Information for further experimental details.

combination with 1,4-benzoquinone<sup>23</sup> delivered the desired product of metathesis 9 in 52% yield. Hydrogenation of the C(25)−C(26) double bond of comp[oun](#page-3-0)d 9 using palladium on carbon resulted in partial hydrogenation of the carbon−iodine bond. Use of Crabtree's catalyst<sup>24</sup> prevented this side reaction but led to complete epimerization of  $C(27)$ . In contrast, diimide-mediated hydrogenatio[n o](#page-3-0)f the C(25)−C(26) double bond occurred smoothly.<sup>25</sup> Subsequent Bernet−Vasella cleav $age<sup>26</sup>$  of the iodoether delivered the terminal olefin 10, which upon TBS-protection foll[ow](#page-3-0)ed by Johnson−Lemieux oxidative cle[ava](#page-3-0)ge provides the aldehyde 11. Similarly, Miyashita's premisakinolide A (bistheonellic acid B) intermediate 12 is prepared through MOM-protection followed by alkene oxidative cleavage in a total of 9 steps (LLS) from  $(S)$ -1,3butane diol 1.

In summary, using technology for the direct stereo- and siteselective conversion of lower alcohols to higher alcohols via C− C bond forming transfer hydrogenation, we report a formal synthesis of premisakinolide A (bistheonellic acid B). The intercepted substructure, which was previously made in 16−27 steps  $(\text{LLS})$ ,  $^{12b,i,15b,h}$  is now made in only 9 steps (LLS). This fragment also represents a synthesis of C(19)−C(32) of the actin-bindin[g macrod](#page-3-0)iolide swinholide A. This study, along with prior work from our laboratory,<sup>16b</sup> highlights the strategic advantages of site-selectivity and redox-economy in chemical synthesis.

# ASSOCIATED CONTENT

#### **S** Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.5b02056.

Spectral data for all new compounds  $(^1H$  NMR,  $^{13}C$ NMR, IR, and HRMS) (PDF)

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#### **Notes**

The authors declare no competing financial interest.

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### ■ REFERENCES

(1) For reviews, see: (a) Jordan, M. A.; Wilson, L. Nat. Rev. Cancer 2004, 4, 253. (b) Dumontet, C.; Jordan, M. A. Nat. Rev. Drug Discovery 2010, 9, 790. (c) Rohena, C. C.; Mooberry, S. L. Nat. Prod. Rep. 2014, 31, 335.

(2) For a review, see: (a) Spector, I.; Braet, F.; Shochet, N. R.; Bubb, M. R. Microsc. Res. Tech. 1999, 47, 18. (b) Yeung, K.-S.; Paterson, I. Angew. Chem., Int. Ed. 2002, 41, 4632. (c) Giganti, A.; Friederich, E. Prog. Cell Cycle Res. 2003, 5, 511. (d) Kita, M.; Kigoshi, H. Nat. Prod. Rep. 2015, 32, 534.

(3) For isolation and structure determination of swinholide A, see: (a) Carmely, S.; Kashman, Y. Tetrahedron Lett. 1985, 26, 511. (b) Kobayashi, M.; Tanaka, J.; Katori, T.; Matsuura, M.; Kitagawa, I. Tetrahedron Lett. 1989, 30, 2963. (c) Kitagawa, I.; Kobayashi, M.; Katori, T.; Yamashita, M.; Tanaka, J.; Doi, M.; Ishida, T. J. Am. Chem. Soc. 1990, 112, 3710. (d) Kobayashi, M.; Tanaka, J.; Katori, T.; Matsuura, M.; Yamashita, M.; Kitagawa, I. Chem. Pharm. Bull. 1990, 38, 2409. (e) Doi, M.; Ishida, T.; Kobayashi, M.; Kitagawa, I. J. Org. Chem. 1991, 56, 3629.

(4) Bewley, C. A.; Holland, N. D.; Faulkner, D. J. Experientia 1996, 52, 716.

(5) Swinholides B−K and ankaraholides A and B: (a) Kobayashi, M.; Tanaka, J.-i.; Katori, T.; Kitagawa, I. Chem. Pharm. Bull. 1990, 38, 2960. (b) Tsukamoto, S.; Ishibashi, M.; Sasaki, T.; Kobayashi, J.-i. J. Chem. Soc., Perkin Trans. 1 1991, 3185. (c) Dumdei, E. J.; Blunt, J. W.; Munro, M. H. G.; Pannell, L. K. J. Org. Chem. 1997, 62, 2636. (d) Youssef, D. T. A.; Mooberry, S. L. J. Nat. Prod. 2006, 69, 154. (e) Marino, S.; Festa, C.; D'Auria, M. V.; Cresteil, T.; Debitus, C.; Zampella, A. Mar. Drugs 2011, 9, 1133. (f) Sinisi, A.; Calcinai, B.; Cerrano, C.; Dien, H. A.; Zampella, A.; D'Amore, C.; Renga, B.; Fiorucci, S.; Taglialatela-Scafati, O. Bioorg. Med. Chem. 2013, 21, 5332. (g) Andrianasolo, E. H.; Gross, H.; Goeger, D.; Musafija-Girt, M.; McPhail, K.; Leal, R. M.; Mooberry, S. L.; Gerwick, W. H. Org. Lett. 2005, 7, 1375.

(6) Todd, J. S.; Alvi, K. A.; Crews, P. Tetrahedron Lett. 1992, 33, 441. (7) (a) Sakai, R.; Higa, T.; Kashman, Y. Chem. Lett. 1986, 1499. (b) Kato, Y.; Fusetani, N.; Matsunaga, S.; Hashimoto, K.; Sakai, R.; Higa, T.; Kashman, Y. Tetrahedron Lett. 1987, 28, 6225. (c) Kobayashi, J.; Tsukamoto, S.; Tanabe, A.; Sasaki, T.; Ishibashi, M. J. Chem. Soc., Perkin Trans. 1 1991, 2379.

(8) (a) Moore, R. E.; Patterson, G. M. L.; Mynderse, J. S.; Barchi, J., Jr.; Norton, T. R.; Furusawa, E.; Furusawa, S. Pure Appl. Chem. 1986, 58, 263. (b) Ishibashi, M.; Moore, R. E.; Patterson, G. M. L.; Xu, C.; Clardy, J. J. Org. Chem. 1986, 51, 5300.

(9) Kobayashi, M.; Kawazoe, K.; Okamoto, T.; Sasaki, T.; Kitagawa, I. Chem. Pharm. Bull. 1994, 42, 19.

<span id="page-3-0"></span>(10) (a) Spector, I.; Bershadsky, A. D.; Korn, E. D. J. Biol. Chem. 1995, 270, 3463. (b) Saito, S.-y.; Karaki, H. Clin. Exp. Pharmacol. Physiol. 1996, 23, 743. (c) Klenchin, V. A.; King, R.; Tanaka, J.; Marriott, G.; Rayment, I. Chem. Biol. 2005, 12, 287.

(11) (a) Perrins, R. D.; Cecere, G.; Paterson, I.; Marriott, G. Chem. Biol. 2008, 15, 287. (b) Herkommer, D.; Dreisigacker, S.; Sergeev, G.; Sasse, F.; Gohlke, H.; Menche, D. ChemMedChem 2015, 10, 470 and references cited therein..

(12) (a) Paterson, I. Pure Appl. Chem. 1992, 64, 1821. (b) Paterson, I.; Cumming, J. G. Tetrahedron Lett. 1992, 33, 2847. (c) Paterson, I.; Smith, J. D. J. Org. Chem. 1992, 57, 3261. (d) Paterson, I.; Smith, J. D. Tetrahedron Lett. 1993, 34, 5351. (e) Paterson, I.; Cumming, J. G.; Smith, J. D.; Ward, R. A. Tetrahedron Lett. 1994, 35, 441. (f) Paterson, I.; Cumming, J. G.; Smith, J. D.; Ward, R. A.; Yeung, K.-S. Tetrahedron Lett. 1994, 35, 3405. (g) Paterson, I.; Smith, J. D.; Ward, R. A.; Cumming, J. G. J. Am. Chem. Soc. 1994, 116, 2615. (h) Paterson, I.; Yeung, K.-S.; Ward, R. A.; Cumming, J. G.; Smith, J. D. J. Am. Chem. Soc. 1994, 116, 9391. (i) Paterson, I.; Cumming, J. G.; Ward, R. A.; Lamboley, S. Tetrahedron 1995, 51, 9393. (j) Paterson, I.; Smith, J. D.; Ward, R. A. Tetrahedron 1995, 51, 9413. (k) Paterson, I.; Ward, R. A.; Smith, J. D.; Cumming, J. G.; Yeung, K.-S. Tetrahedron 1995, 51, 9437. (l) Paterson, I.; Yeung, K.-S.; Ward, R. A.; Smith, J. D.; Cumming, J. G.; Lamboley, S. Tetrahedron 1995, 51, 9467.

(13) (a) Patron, A. P.; Richter, P. K.; Tomaszewski, M. J.; Miller, R. A.; Nicolaou, K. C. J. Chem. Soc., Chem. Commun. 1994, 1147. (b) Richter, P. K.; Tomaszewski, M. J.; Miller, R. A.; Patron, A. P.; Nicolaou, K. C. J. Chem. Soc., Chem. Commun. 1994, 1151. (c) Nicolaou, K. C.; Patron, A. P.; Ajito, K.; Richter, P. K.; Khatuya, H.; Bertinato, P.; Miller, R. A.; Tomaszewski, M. J. Chem. - Eur. J. 1996, 2, 847. (d) Nicolaou, K. C.; Ajito, K.; Patron, A. P.; Khatuya, H.; Richter, P. K.; Bertinato, P. J. Am. Chem. Soc. 1996, 118, 3059.

(14) (a) Nakata, T.; Komatsu, T.; Nagasawa, K. Chem. Pharm. Bull. 1994, 42, 2403. (b) Nakata, T.; Komatsu, T.; Nagasawa, K.; Yamada, H.; Takahashi, T. Tetrahedron Lett. 1994, 35, 8225. (c) Nagasawa, K.; Shimizu, I.; Nakata, T. Tetrahedron Lett. 1996, 37, 6881. (d) Nagasawa, K.; Shimizu, I.; Nakata, T. Tetrahedron Lett. 1996, 37, 6885.

(15) (a) Mulzer, J.; Meyer, F.; Buschmann, J.; Luger, P. Tetrahedron Lett. 1995, 36, 3503. (b) Keck, G. E.; Lundquist, G. D. J. Org. Chem. 1999, 64, 4482. (c) Kartika, R.; Frein, J. D.; Taylor, R. E. J. Org. Chem. 2008, 73, 5592. (d) Hayakawa, H.; Miyashita, M. J. Chem. Soc., Perkin Trans. 1 1999, 3399. (e) Hayakawa, H.; Iida, K.; Miyazawa, M.; Miyashita, M. Chem. Lett. 1999, 601. (f) Hayakawa, H.; Miyashita, M. Tetrahedron Lett. 2000, 41, 707. (g) Nakamura, R.; Tanino, K.; Miyashita, M. Org. Lett. 2003, 5, 3579. (h) Nakamura, R.; Tanino, K.; Miyashita, M. Org. Lett. 2005, 7, 2929.

(16) Selected reviews: (a) Ketcham, J. M.; Shin, I.; Montgomery, T. P.; Krische, M. J. Angew. Chem., Int. Ed. 2014, 53, 9142. (b) Dechert-Schmitt, A.-M. R.; Schmitt, D. C.; Gao, X.; Itoh, T.; Krische, M. J. Nat. Prod. Rep. 2014, 31, 504.

(17) Shin, I.; Wang, G.; Krische, M. J. Chem. - Eur. J. 2014, 20, 13382. (18) (a) Gao, X.; Han, H.; Krische, M. J. J. Am. Chem. Soc. 2011, 133, 12795. (b) Gao, X.; Woo, S. K.; Krische, M. J. J. Am. Chem. Soc. 2013,

135, 4223. (19) For related cross-metatheses, see: (a) Lucas, B. S.; Burke, S. D. Org. Lett. 2003, 5, 3915. (b) Lee, K.; Kim, H.; Hong, J. Org. Lett. 2009, 11, 5202.

(20) (a) Vigneron, J. P.; Dhaenens, M.; Horeau, A. Tetrahedron 1973, 29, 1055. (b) For a historical review, see: Heller, D.; Drexler, H.-J.; Fischer, C.; Buschmann, H.; Baumann, W.; Heller, B. Angew. Chem., Int. Ed. 2000, 39, 495.

(21) (a) Kogure, T.; Eliel, E. L. J. Org. Chem. 1984, 49, 576. (b) Midland, M. M.; Gabriel, J. J. Org. Chem. 1985, 50, 1143. (c) Poss, C. S.; Schreiber, S. L. Acc. Chem. Res. 1994, 27, 9.

(22) Ghosh, A. K.; Cheng, X.; Bal, R.; Hamel, E. Eur. J. Org. Chem. 2012, 2012, 4130.

(23) Hong, S. H.; Sanders, D. P.; Lee, C. W.; Grubbs, R. H. J. Am. Chem. Soc. 2005, 127, 17160.

(24) Review: Brown, J. M. Angew. Chem., Int. Ed. Engl. 1987, 26, 190.

(25) (a) Dewey, R. S.; van Tamelen, E. E. J. Am. Chem. Soc. 1961, 83, 3729. (b) Zhu, K.; Panek, J. S. Org. Lett. 2011, 13, 4652. (c) Cheung, L. L.; Marumoto, S.; Anderson, C. D.; Rychnovsky, S. D. Org. Lett. 2008, 10, 3101. (d) Kang, E. J.; Cho, E. J.; Ji, M. K.; Lee, Y. E.; Shin, D. M.; Choi, S. Y.; Chung, Y. K.; Kim, J.-S.; Kim, H.-J.; Lee, S.-G.; Lah, M. S.; Lee, E. J. Org. Chem. 2005, 70, 6321.

(26) (a) Bernet, B.; Vasella, A. Helv. Chim. Acta 1979, 62, 1990. (b) Bernet, B.; Vasella, A. Helv. Chim. Acta 1979, 62, 2400. (c) Bernet, B.; Vasella, A. Helv. Chim. Acta 1984, 67, 1328.